

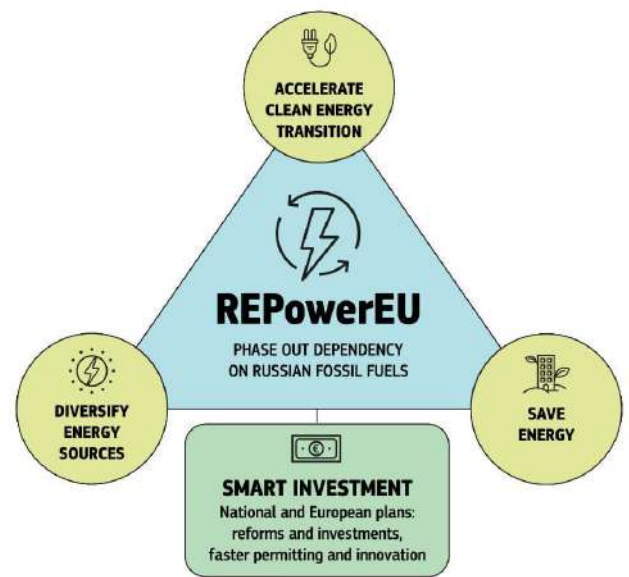
Valorization of Sludge Digestate and Biomass Wastes for the Electro-Generation of Hydrogen Peroxide

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REPowerEU plan 2022



Reduce fossil fuel dependency Ecological transition

Biomethane 35 billion cubic meters by 2030

Digestate generation 180 million metric tons annually

Introduction

Fertilizer for soil

Energy source

Environmental applications

Electrochemistry



Electrochemistry

Fuel cells

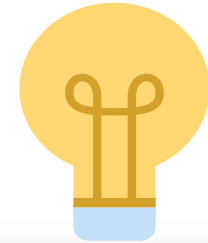
MFCs

Supercapacitors

Electrochemical sensors



Digestate

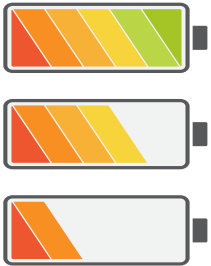


Limited research

Introduction



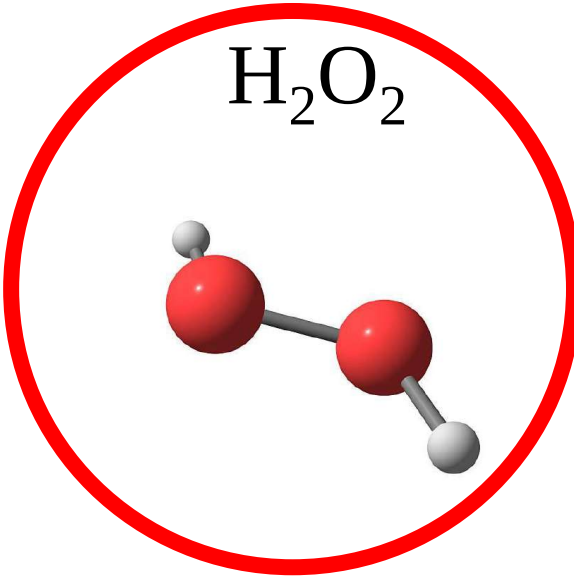
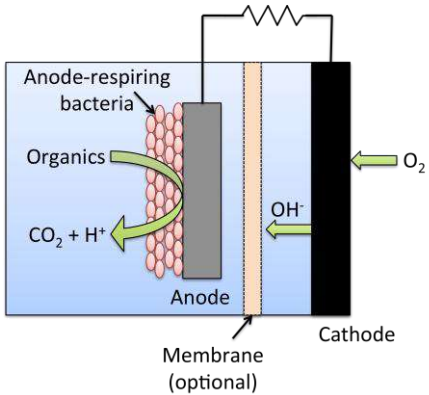
Energy storage



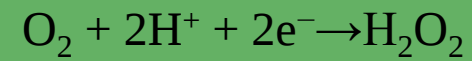
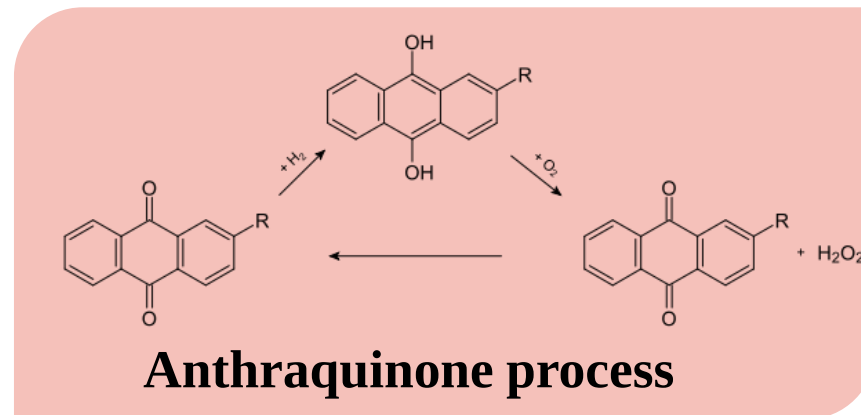
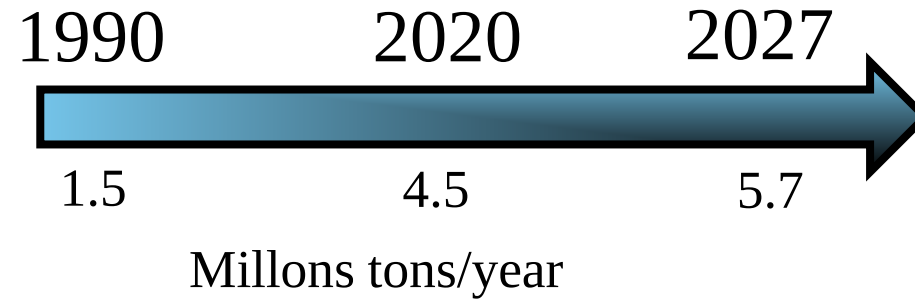
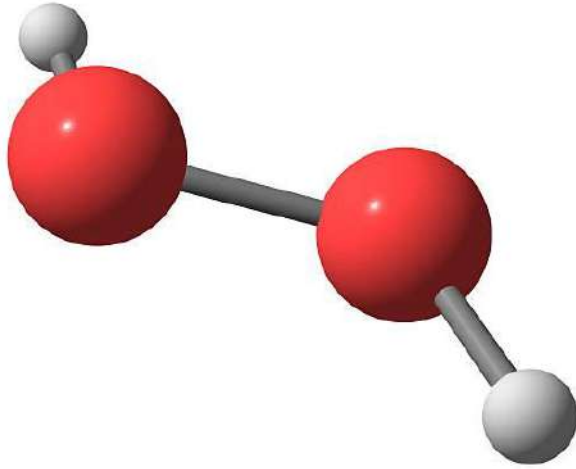
Hydrogen electrogeneration



MFCs



Introduction



Electrochemical generation

Introduction



Valorization of polluted biomass waste for manufacturing sustainable cathode materials for the production of hydrogen peroxide

Álvaro Ramírez, Martín Muñoz-Morales, Francisco Jesús Fernández-Morales, Javier Llanos*

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Spargularia rubra



Innovative carbon materials from lignocellulosic wastes for electrochemical hydrogen peroxide production: Bridging biomass conversion and material properties

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Phragmites australis



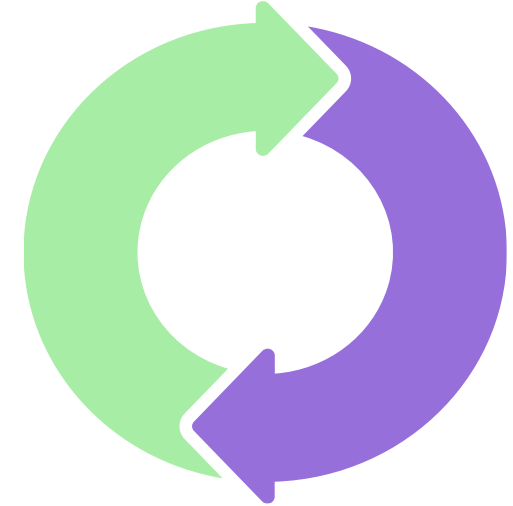
Cladium mariscus



Typha domingensis

Objective

Study the capacity of digestate as a catalyst for the electrogeneration of H_2O_2



Digestate



+



Digestate + *Phragmites australis*

Methodology



Hydrothermal carbonization (HTC)

200 °C / 2 h

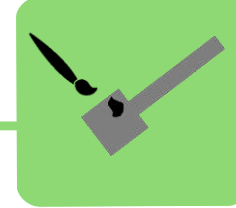
- **D200**
- **PhA200**
- **H1:1**



Pyrolysis and chemical activation

1000 °C / 600 °C KOH

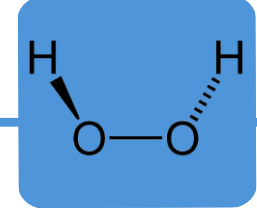
- **D200-KOH**
- **H1:1-KOH**
- **H1:1-1000**



Electrode deposition

Immersion method

- Ultrasonic bath



Electrochemical generation of H₂O₂

- 2 h accumulation tests
- -0.9 V vs Ag/AgCl
- Aeration + stirring
- Undivided cell

Physicochemical characterization

EDS

FTIR

Raman

N₂
adsorption
isotherms

HRSEM

Electrochemical characterization

Linear sweep
voltammetry
(LSV)

Chronoamperometry
(CA)

Results

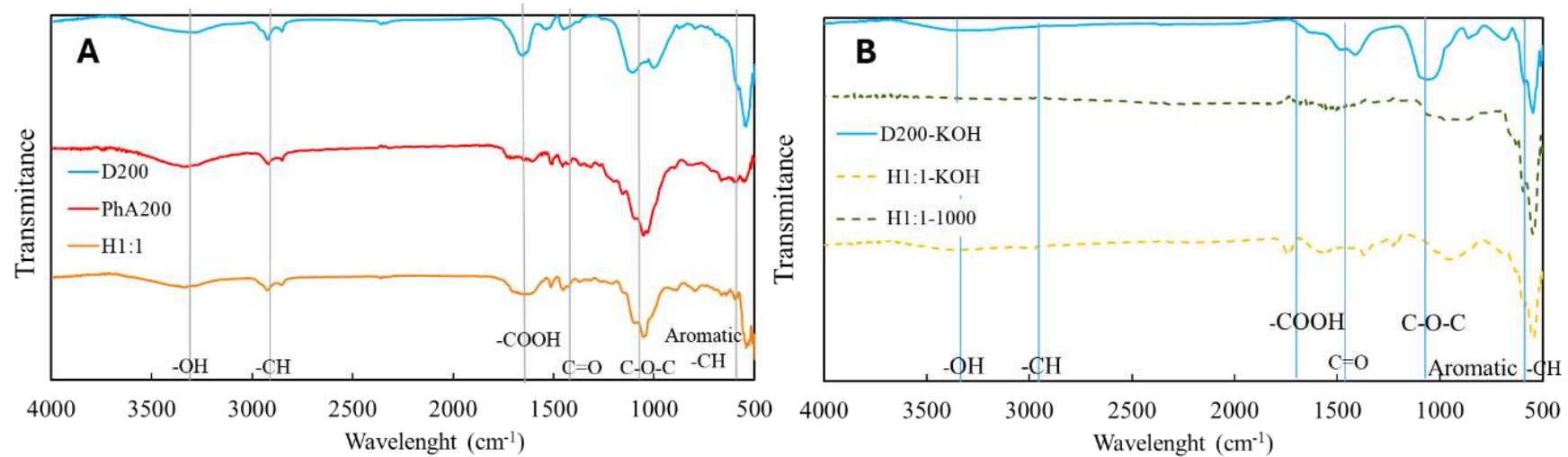
EDS

Parameters	PhA200	D200	H1:1	D200-KOH	H1:1-KOH	H1:1- 1000
C (% at)	70.73	52.97	67.10	38.63	53.89	76.72
O (% at)	29.22	33.62	24.93	36.23	26.60	12.78
Na (% at)	-	-	-	-	0.12	0.11
Mg (% at)	-	-	0.38	1.81	0.97	0.57
Al (% at)	0.02	0.89	0.56	2.06	1.98	0.66
Si (%at)	0.03	1.56	2.36	4.1	6.3	3.5
P (% at)	-	4.00	1.62	2.6	0.4	1.19
S (% at)	-	0.42	0.30	0.64	1.06	0.36
K (% at)	-	0.1	0.08	2.93	2.28	0.45
Ca (% at)	-	2.9	1.18	5.37	2.22	1.65
Ti (% at)	-	0.09	-	0.36	0.29	0.1
Fe (% at)	-	3.45	1.49	5.27	3.89	1.91

- ✓ Differences in C content
- ✓ Presence of Fe in samples with digestate

Results

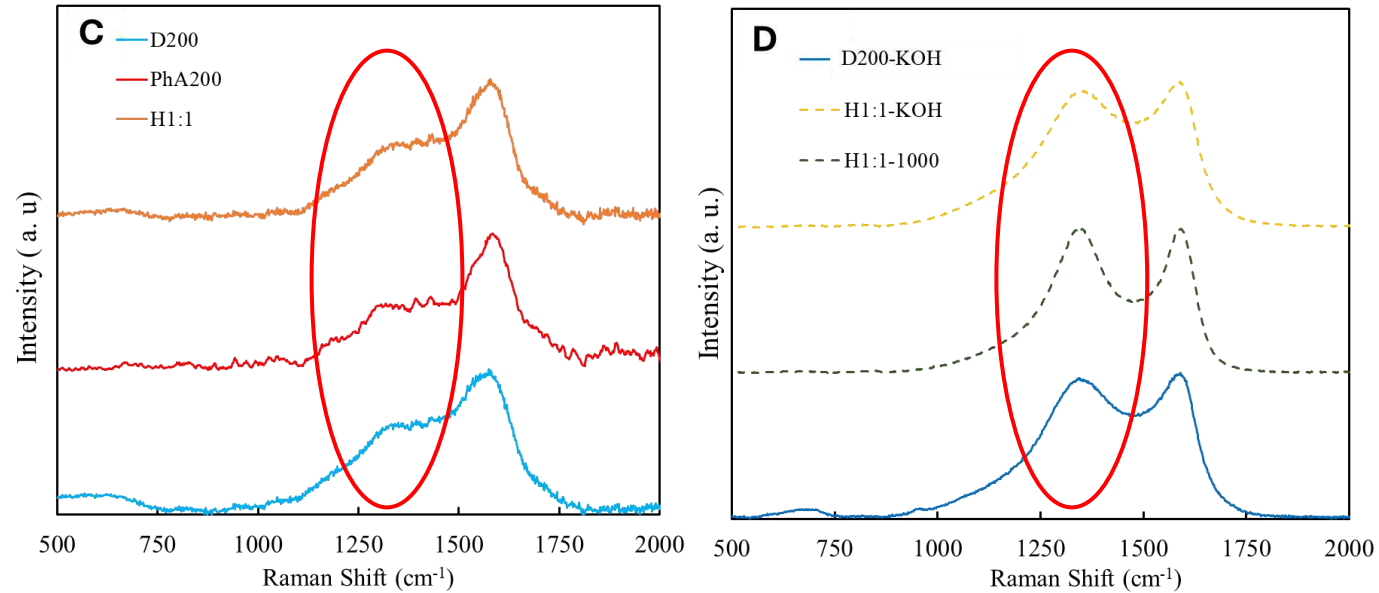
FTIR



- ✓ Reduction of functional groups as the treatment temperature increases

Results

Raman

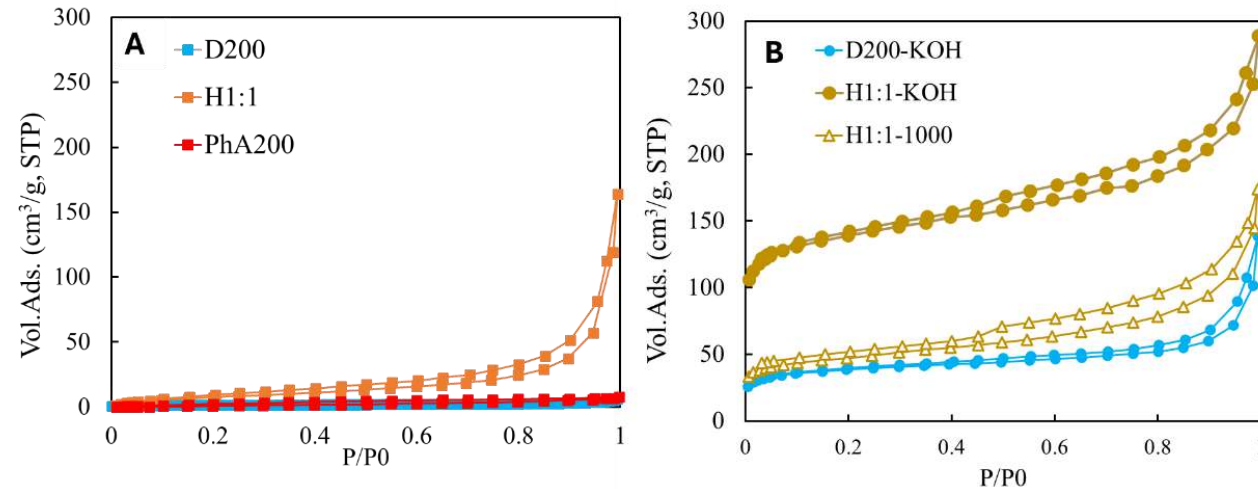


	D200	D200-KOH	PhA200	H1:1	H1:1-KOH	H1:1-1000
AD_1/AG	0.81	1.52	0.63	0.75	1.53	1.44

- ✓ Absent of possible defects and lack of sp^2 graphitic network ($AD_1/AG < 1$)
- ✓ D-band increase due to the growth of sp^2 domains in activated samples enhancing the defects ($AD_1/AG \approx 1.5$)

Results

N₂ adsorption isotherms

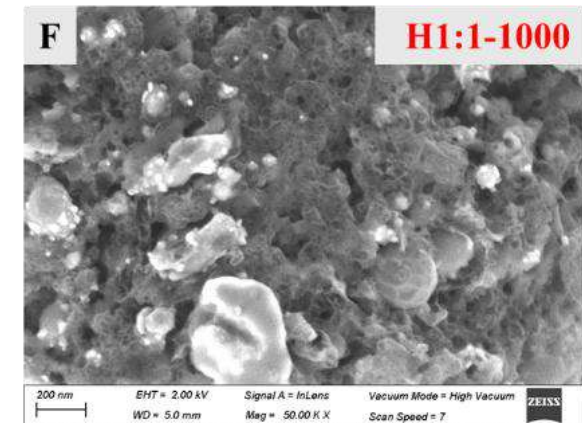
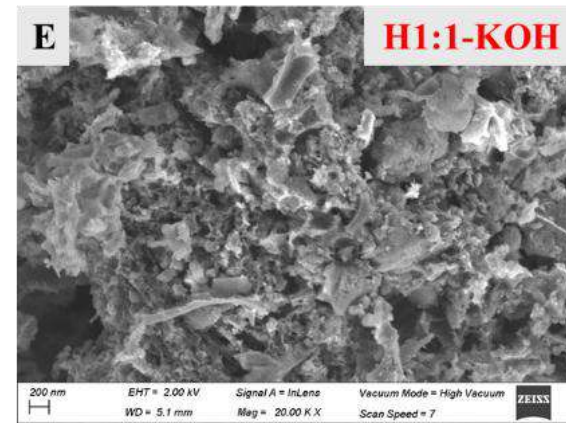
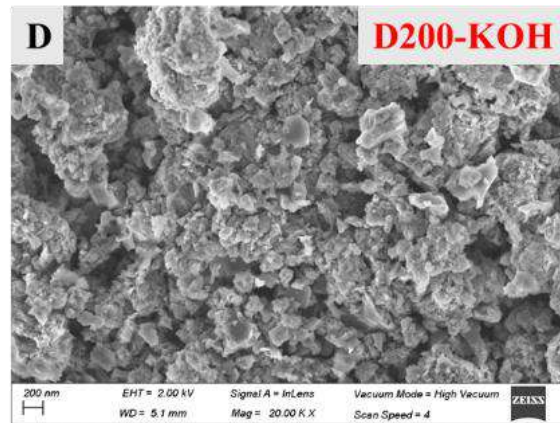
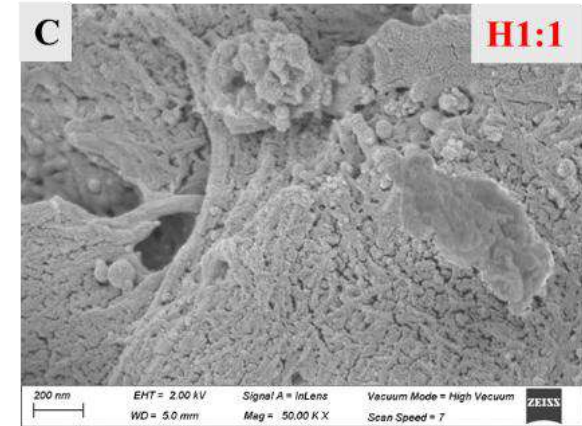
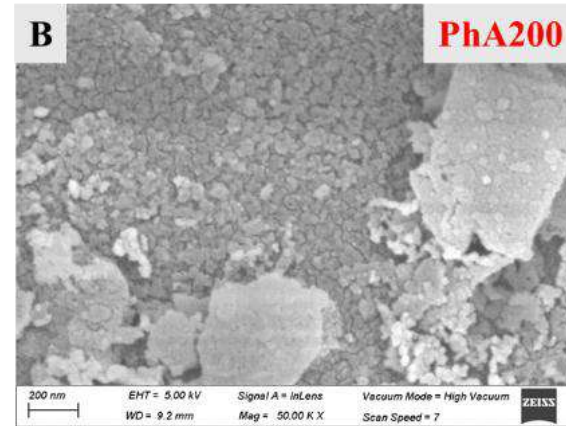
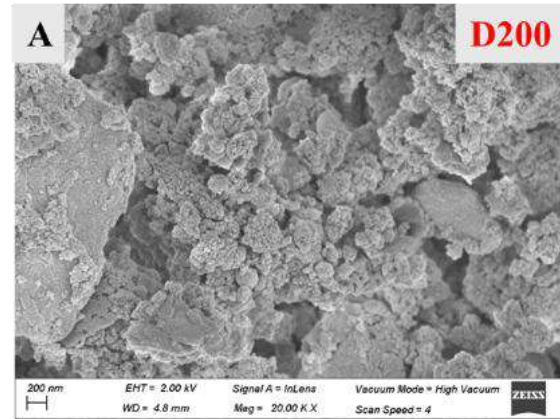


	D200	D200-KOH	PhA200	H1:1	H1:1-KOH	H1:1-1000
Specific Surface area (m ² /g)	5	142	14	31	525	174

- ✓ Lower surface area when chemically activating the digestate compared to *Phragmites australis*
- ✓ Higher thermal stability for digestate (inorganic compounds in its structure)

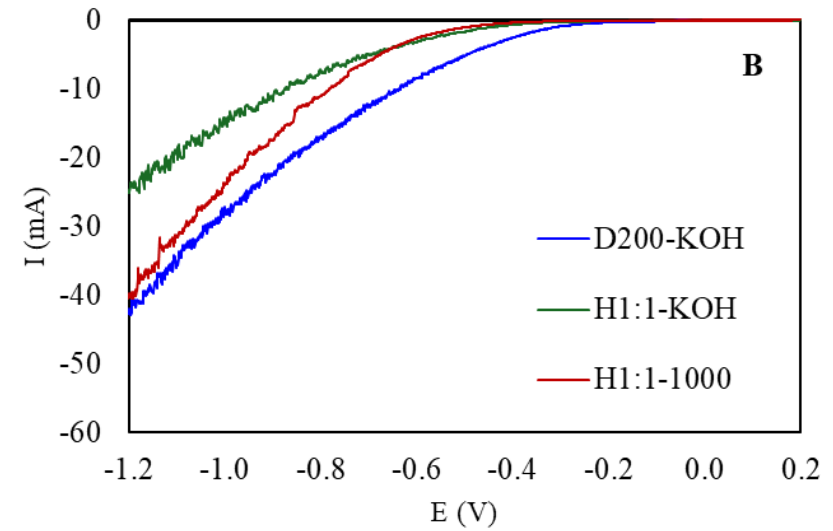
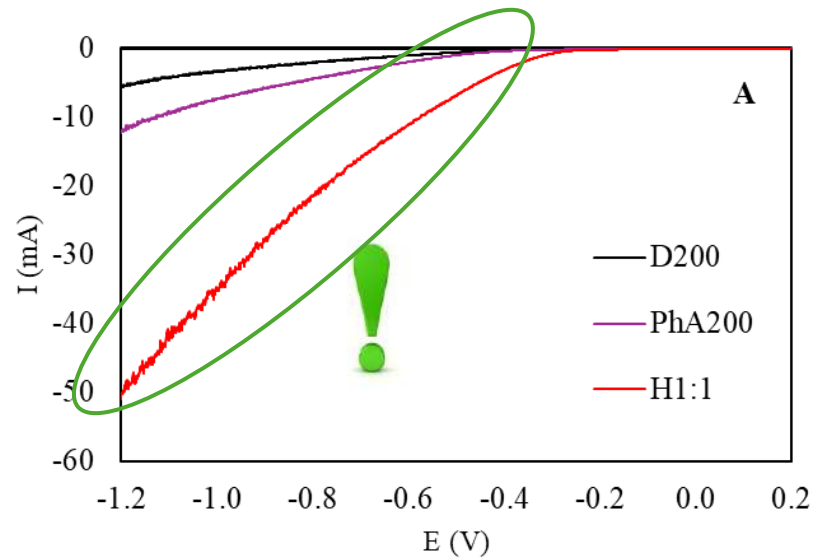
Results

HRSEM



Results

LSV

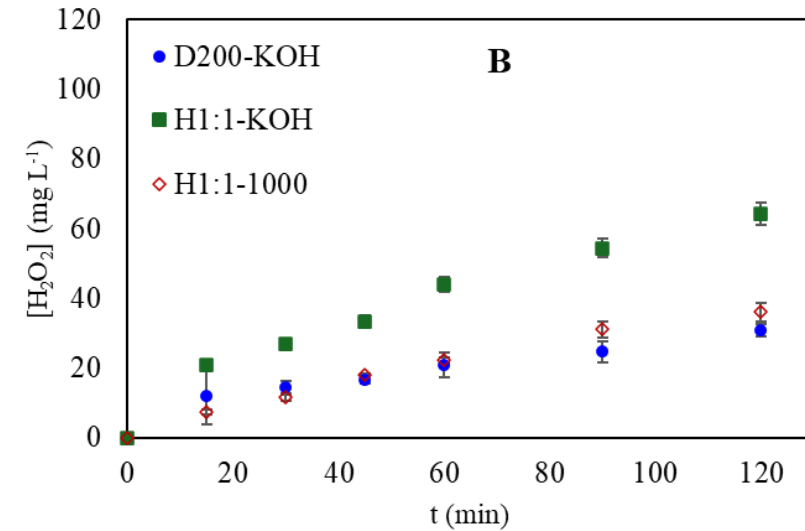
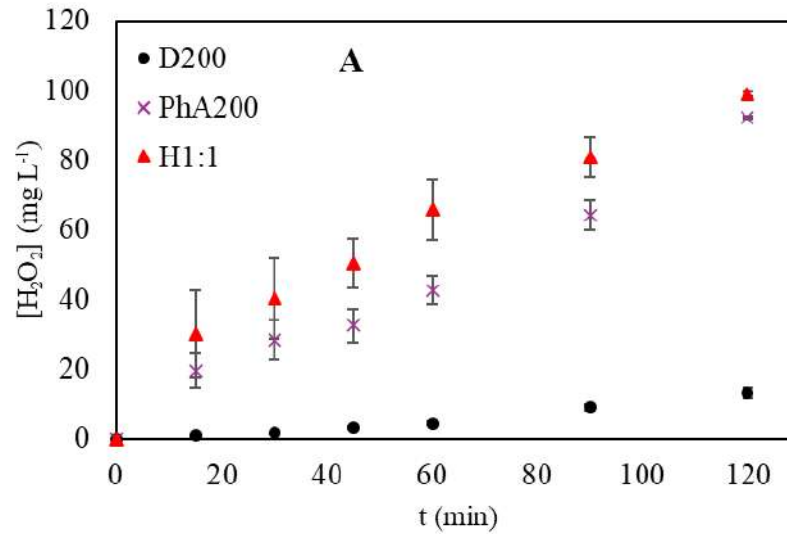


- ✓ Synergy between digestate and *Phragmites australis* in H1:1
- ✓ Lower intensities for the activates samples

Results

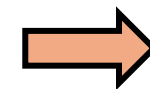
CA

- 9 cm² cathodes
- -0.9 V vs Ag/AgCl
- Undivided cell
- Constant aeration and stirring
- 0.05 M Na₂SO₄ electrolyte



✓ Poor results from D200. Good results when combined with *Phragmites australis* (H1:1)

✓ Better results for the hydrochars than for the activated samples



Presence of Fe in the digestate

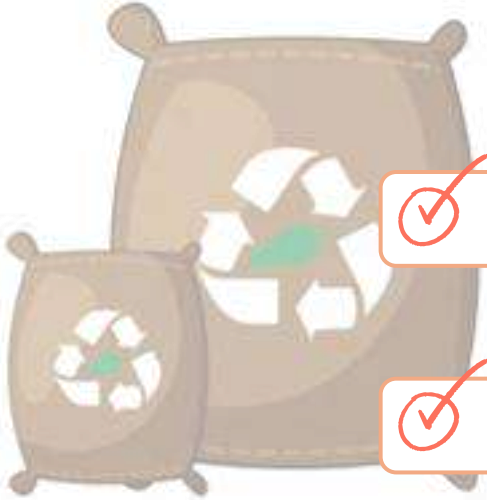
Results

CA

- 9 cm² cathodes
- -0.9 V vs Ag/AgCl
- Undivided cell
- Constant aeration and stirring
- 0.05 M Na₂SO₄ electrolyte

	FE (%)	EC (kWh kgH ₂ O ₂ ⁻¹)
PhA200	100	4.2
H1:1	72	5.6
D200-KOH	56	7.0
H1:1-KOH	59	6.6
H1:1-1000	30	11.5

Conclusions



Feasibility of using digestate for synthesizing carbonaceous catalysts



Better performance of the digestate-*Phragmites australis* mixture Hydrochar (H1:1)



Worst behaviour from activated materials



Possibility of studying new digestate-*Phragmites australis* ratios



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THANK YOU

